



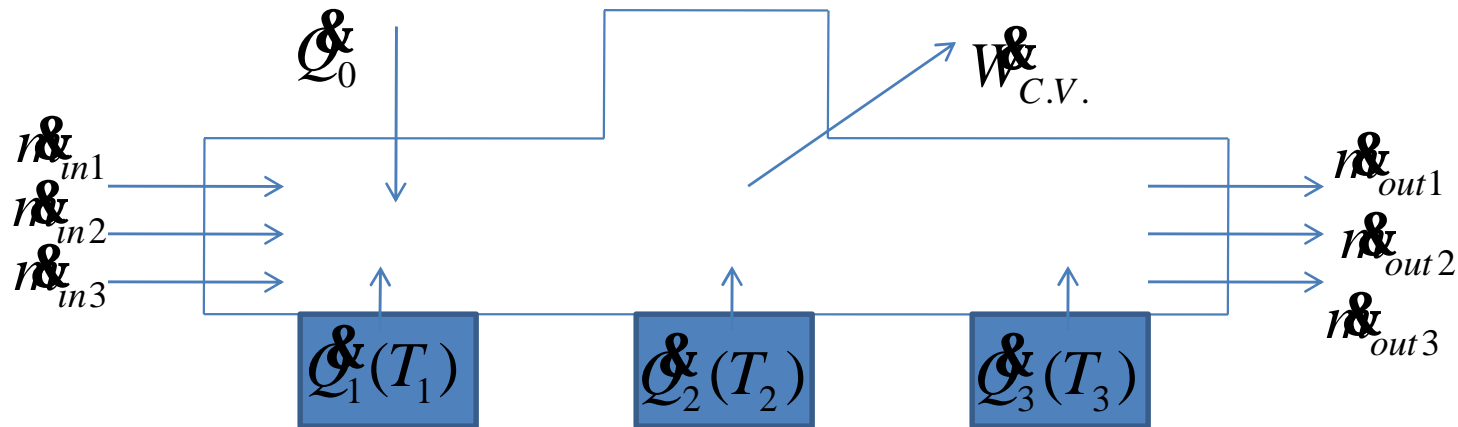
**Sharif University of Technology  
School of Mechanical Engineering  
Center of Excellence in Energy Conversion**

# **Advanced Thermodynamics**

## **Lecture 12**

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∅ Thermodynamic analysis

$$\frac{dm_{C.V.}}{dt} = \sum_{in} \dot{m} - \sum_{out} \dot{m} \Rightarrow \Delta_{1-2} m_{C.V.} = \sum \int_{t_1}^{t_2} \dot{m}_{in} dt - \sum \int_{t_1}^{t_2} \dot{m}_{out} dt$$

∅ Energy Eq. (Heat balance) or 1<sup>st</sup> law  $(h^* = h + \frac{v^2}{2} + gz)$

$$\frac{dE_{C.V.}}{dt} = \sum_{i=0}^n \dot{Q}_i - \dot{W}_{C.V.} + \sum_{in} \dot{m} h^* - \sum_{out} \dot{m} h^* \quad \text{Eq. (1)}$$

∅ Entropy principle (2<sup>nd</sup> law)

$$\frac{dS_{C.V.}}{dt} = \sum_{i=0}^n \frac{\dot{Q}_i}{T_i} + \sum_{in} \dot{m} S - \sum_{out} \dot{m} S + \dot{S}_{gen} \quad \text{Eq. (2)}$$

∅ From Eq. (1) 
$$\dot{Q}_0 = \frac{dE_{C.V.}}{dt} + \sum_{i=1}^n \dot{Q}_i - \dot{W}_{C.V.} + \sum_{in} \dot{m} h^* - \sum_{out} \dot{m} h^*$$

∅ From Eq. (2)

$$\dot{Q}_0 = T_0 \frac{dS_{C.V.}}{dt} - T_0 \sum_{i=0}^n \frac{\dot{Q}_i}{T_i} - T_0 \sum_{in} \dot{m} S + T_0 \sum_{out} \dot{m} S - T_0 \dot{S}_{gen}$$

$$\Rightarrow \frac{dE_{C.V.}}{dt} + P_0 \frac{dV}{dt} - T_0 \frac{dS_{C.V.}}{dt} = \sum_{i=1}^n \left( 1 - \frac{T_0}{T} \right) \dot{Q}_i - \left( \dot{W}_{C.V.} - P_0 \frac{dV}{dt} \right) + \sum_{in} \dot{m} (h^* - T_0 S) - \sum_{out} \dot{m} (h^* - T_0 S) - T_0 \dot{S}_{gen}$$

∅ Since and are considered constant

$$\frac{d}{dt} [E_{C.V.} + P_0 dV - T_0 S_{C.V.}] = \frac{d(Exergy)}{dt} = \frac{d(Availability)}{dt} = \frac{dy}{dt}$$

$$\frac{dy}{dt} = \sum_{i=1}^n \left( 1 - \frac{T_0}{T} \right) \dot{Q}_i - \left( \dot{W}_{C.V.} - P_0 \frac{dV}{dt} \right) + \sum_{in} \dot{m} b - \sum_{out} \dot{m} b - T_0 \dot{S}_{gen}$$

- Ø Where  $b = (h^* - T_0 S)$
- Ø  $I$ : Rate of change of exergy (availability) of C.V. with the time.
- Ø  $II$ : Exergy transferred into C.V. with heat, note that
- Ø  $III$ : Rate of available work produced by the C.V.
- Ø  $IV$ : Net availability convected into the C.V. with mass flow, also called physical exergy.
- Ø  $V$ : Exergy destruction due to internal irreversibility.

$$\frac{dy}{dt} = \sum_{i=1}^n \left( 1 - \frac{T_0}{T} \right) \dot{Q}_i - \left( \dot{W}_{C.V.} - P_0 \frac{dV}{dt} \right) + \sum_{in} \dot{m} - \sum_{out} \dot{m} - T_0 \dot{S}_{gen}$$

Ø Observations:

Ø Exergy is simply a combination of the 1<sup>st</sup> and 2<sup>nd</sup> laws.

Ø exergy (availability) is completely rigorous, there are new assumptions in deriving its statement.